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(EXO)PLANETARY ATMOSPHERES CHEMICAL MODELS



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Observatoire Aquitain







Physical/chemical diversity of (exo)planetary atmospheres 10 000 1 000 100 Mass (M_{Earth}) 10 0, I 0,01 10⁻¹ 100 101 10² Equivalent semi-major axis (a.u.)



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Physical/chemical diversity of (exo)planetary atmospheres 10 000 Thermodynamic reaction control 1 000 100 Mass (M_{Earth}) 10 0, I **Kinetic** reaction control 0,01 10-1 100 101 10² Equivalent semi-major axis (a.u.)



Physical/chemical diversity of (exo)planetary atmospheres 10 000 **Chemical equilibrium / Reversibility** 1 000 100 Mass (M_{Earth}) 10 0, I **Kinetic constants / Irreversibility** 0,01 10-1 101 10² Equivalent semi-major axis (a.u.)



Physical/chemical diversity of (exo)planetary atmospheres 10 000 Equilibrium models 100 Mass (M_{Earth}) **Terra** incognita 10 0, I **Photochemical** models 0,01 10-1 100 10¹ 102 Equivalent semi-major axis (a.u.)



Need for a new generation of tools and databases for the physical/chemical study of (exo)planetary atmospheres



(Exo)planetary atmospheres chemical models Eric Hébrard (LAB) **Radiative transfert** Dynamics (Photo)chemistry 10 000 1 000 Saturn 100 HD 209458b, HD 189733b Mass (M_{Earth}) Titan Gliese 581d 10 <u>Neptune</u> 0 **Primitive Earth** <u>Mars</u> 0, I 0,01 10-1 100 101 10² Equivalent semi-major axis (a.u.)



ID photochemical modeling of planetary atmospheres

 Photochemical models of interstellar or planetary atmospheres are complex ([0-3]D chemical-dynamical codes with thousands of highly coupled nonlinear equations)

The chemical equations are based on empirical parameters :

$$AB \xrightarrow{h\nu} A + B$$

$$\sigma_i(\lambda, T) \quad q_{i,j}(\lambda, T)$$

 $A + B \xrightarrow{(+M)} C + D$ $k_i(T) = \alpha_i (\frac{T}{300})^{\beta_i} \exp(-\frac{\gamma_i}{T})$

Photodissociations

Neutral-neutral reactions

These empirical parameters are obtained from experiments, calculations and/or **extrapolations** :

They are always evaluated with **[[very] large] uncertainty**

In some conditions, estimated parameters are numerous

Ex : in Titan photochemical models, less than 10% of reaction rates are measured at relevant temperatures.



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Photodissociations

Neutral-neutral reactions

These empirical parameters are obtained from experiments, calculations and/or [more or less [but most often less]] educated-guessed estimations :

They are always evaluated with [[very] large] uncertainty

Most of the cases, **extrapolations** of these parameters are mandatory

Ex : in Titan photochemical models, less than 10% of reaction rates are measured at relevant temperatures.





ID photochemical modeling of planetary atmospheres

































Evaluation and extrapolation of the uncertainties of (photo)chemical parameters







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$$A + B \xrightarrow{(+M)} C + D$$

$$k_i(T) = \alpha_i \left(\frac{T}{300}\right)^{\beta_i} \exp\left(-\frac{\gamma_i}{T}\right)$$

$$F_{k_i(T)} = \sqrt{\sigma_{\alpha_i}^2 F_{k} \sigma_{\beta_i}^2} \ln^2 T + \sigma_{\gamma_i}^2 T^{-2}$$





Evaluation and extrapolation of the uncertainties of (photo)chemical parameters



Temperature (K)



PhotoChemistry In the <u>upper/cold</u> atmospheres of the Solar System, chemical reactions are initiated by radicals UV-induced production.

 $A + hv \longrightarrow A^*$

Only exothermic reactions are included.

 $A + B \rightleftharpoons C + D$

Photochemical networks are wrong at high temperatures

ThermoChemistry

In the <u>deep/hot</u> layers of the giant planet atmospheres of the Solar System, endothermic reactions take place. There is no UV photon.

 $A + B \rightleftharpoons C + D$

Thermochemical networks are unable to take into account non-equilibrium processes (photodissociations, circulation)



Photo/Thermochemistry

In hot dense atmospheres subjected to high UV fluxes, a new situation arises, in which both photodissociations and endothermic reactions matter:

 $A + h\nu \longrightarrow A^*$

 $A + B \rightleftharpoons C + D$



Photothermochemical networks evolve towards thermodynamic equilibrium when photodissociations and circulation are turned off.

All reactions must be balanced

$$\mathbf{A} + \mathbf{B} \underset{\mathbf{k}_{-1}(\mathbf{T})}{\overset{\mathbf{k}_{1}(\mathbf{T})}{\rightleftharpoons} \mathbf{C} + \mathbf{D} } \begin{cases} k_{1}(T) = \alpha_{1}(\frac{T}{300})^{\beta_{1}} \exp(-\frac{\gamma_{1}}{T}) \\ K_{eq} = \frac{k_{1}(T)}{k_{-1}(T)} = \exp\left(-\frac{\Delta_{r}G_{T}^{0}}{RT}\right) \end{cases}$$

Zahnle et al., 2009; 2010 Line et al., 2010 Moses et al., 2011







Hot Jupiter Photo/Thermochemical modeling



Perfectly Stirred Reactor

Shock Tube

Flame speeds

Rapid Compression Machine

Hot Jupiter Photo/Thermochemical modeling

 $\begin{array}{c} \operatorname{AB} \xrightarrow{h\nu} \operatorname{A} + \operatorname{B} \\ \sigma_i(\lambda, T) \quad q_{i,j}(\lambda, T) \end{array}$

Need for high temperature molecular photoabsorption cross-sections

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http://kida.obs.u-bordeauxl.fr/

Poster 3.105 - Valentine Wakelam (LAB/OASU)

