

Time-dependent Anion Chemistry in the CSE IRC+10216

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Why study anions ? (I)

They are good for our health!



Taroko , Taiwan

Why study IS anions ? (II)

- The presence of C-chain anions in interstellar space was predicted long ago on theoretical grounds (e.g. Herbst 1981)
- The first IS anion, C_6H^- , was identified in 2006 (McCarthy et al.)
- These anions are believed to be mainly formed by radiative attachment of an electron on a C-chain radical:



- The observation of IS anions offers the opportunity to test our understanding of radiative attachment reactions

IRC+10216 (CW Leo) as chemical and spectroscopic laboratory

- **78 molecular species** detected to date
- Wealth of **linear C-chain** molecules and **radicals**
- All known **IS anions** observed there
- (Relatively) well constrained physical conditions
- **Expanding envelope** gives access to time dimension

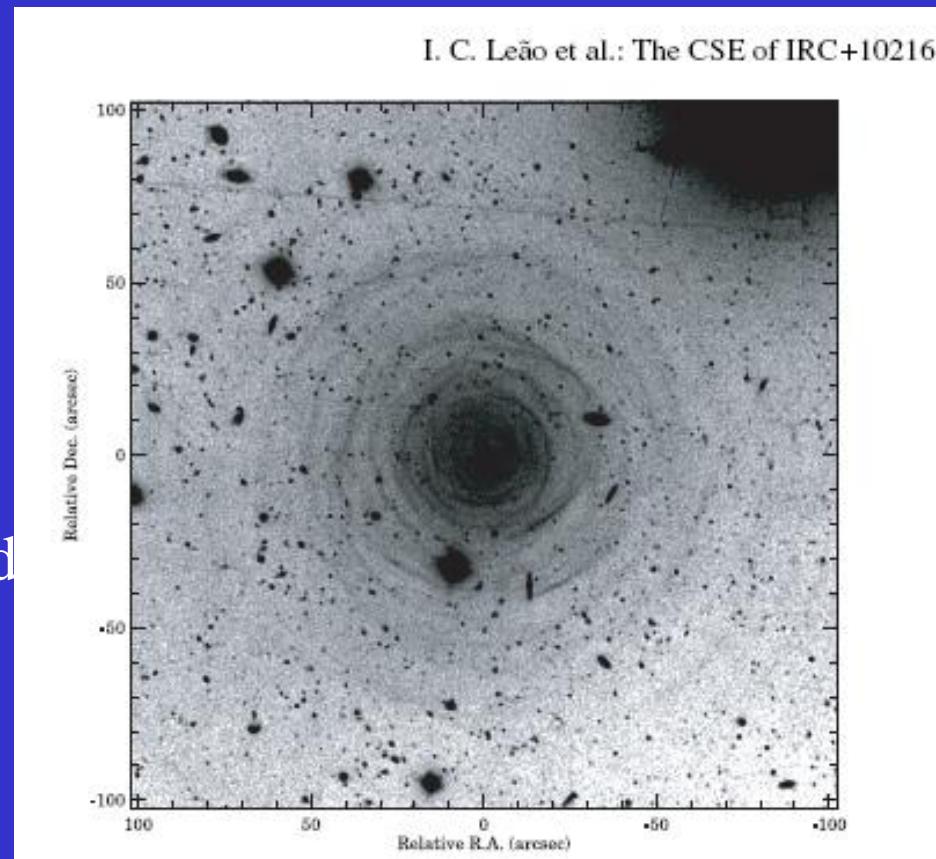


Fig. 3. FORS1 deconvolved V-band image of IRC+10216. North is up and East is left.

1 arcsec ~ 150 A.U.

VLT –V-band optical image

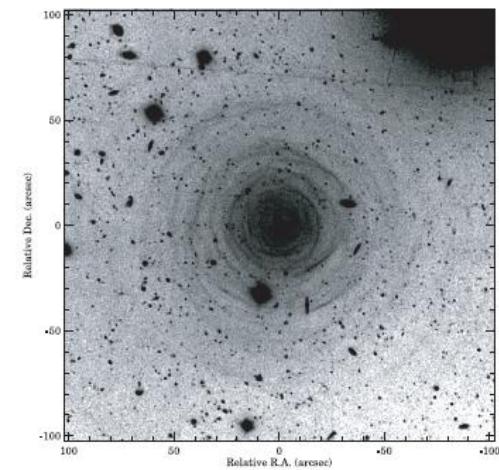
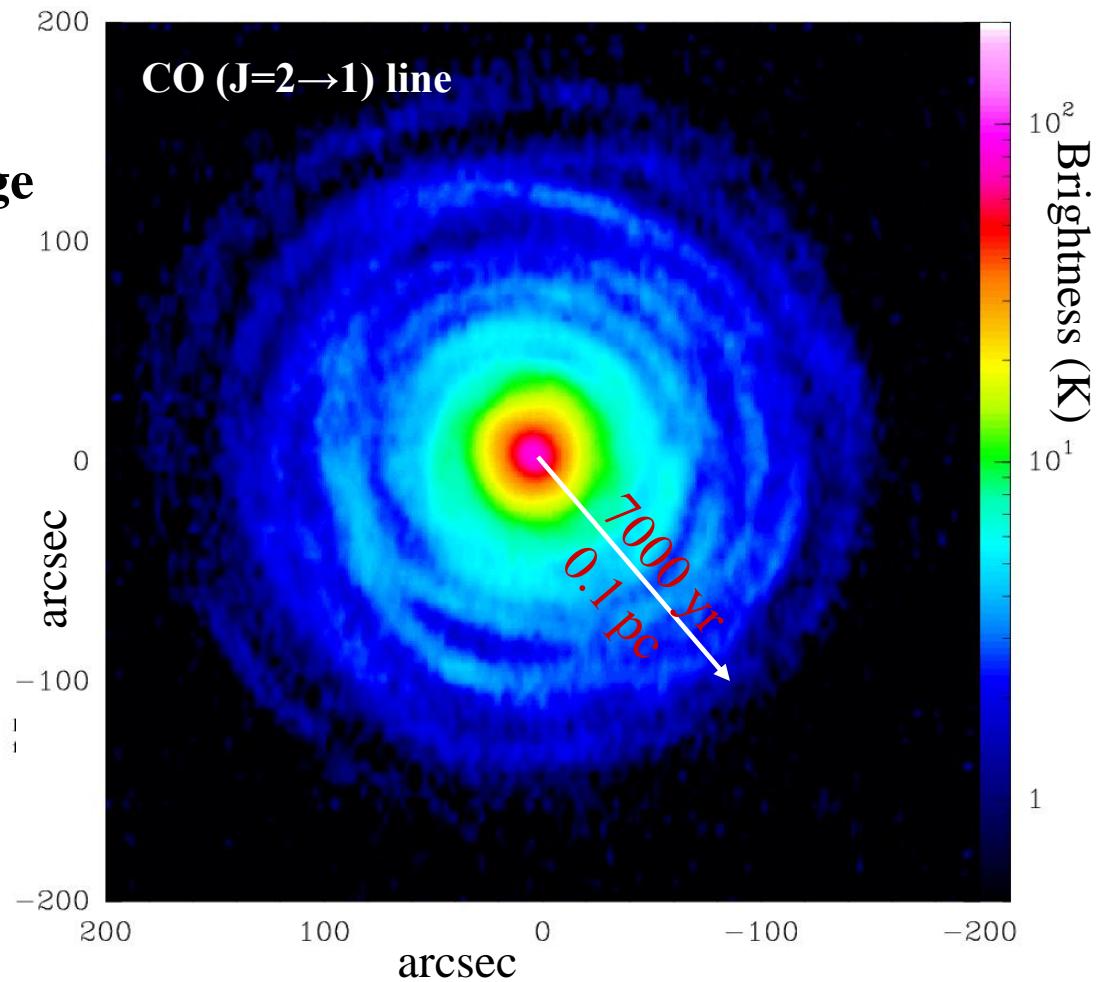
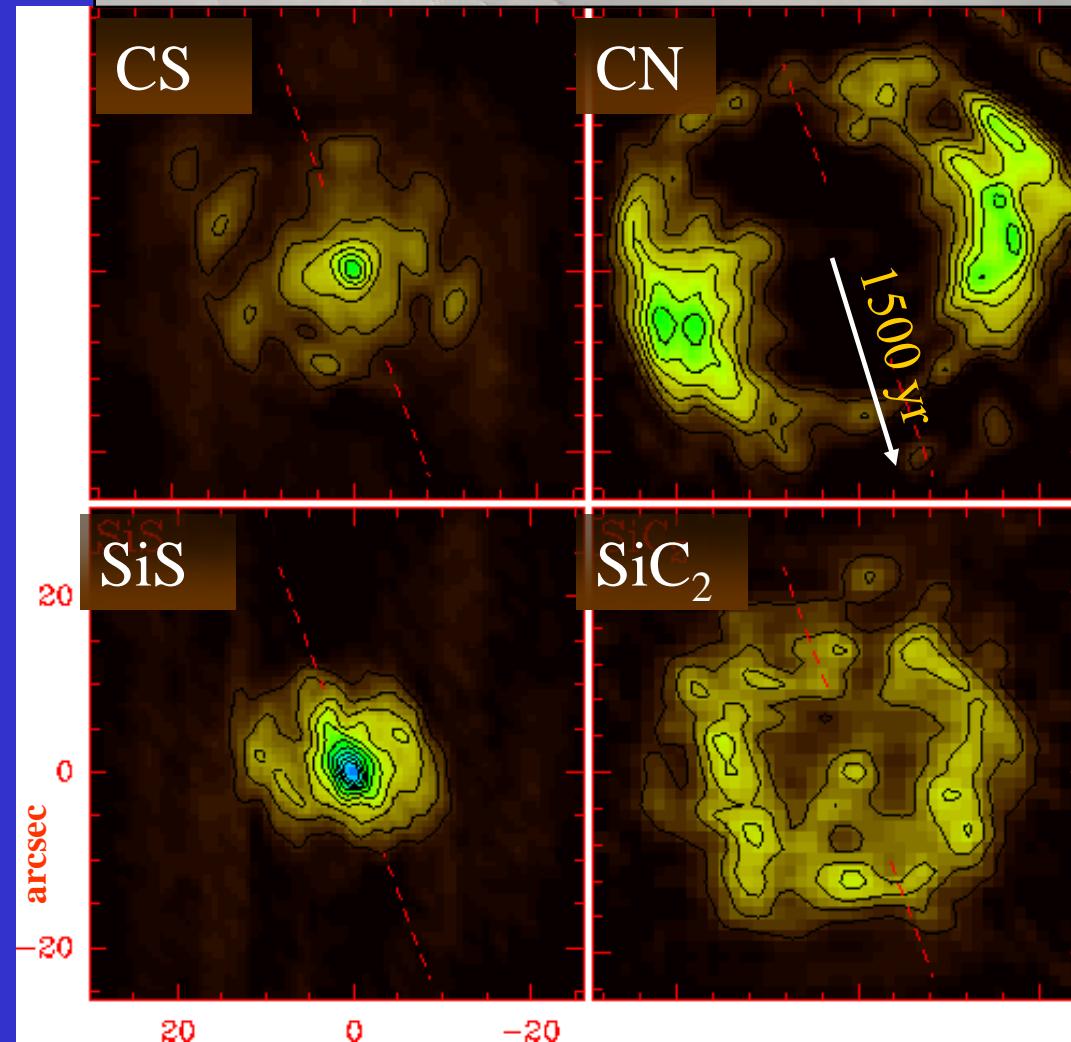


Fig. 3. FORS1 deconvolved V -band image of IRC+10216. North is up and East is left.

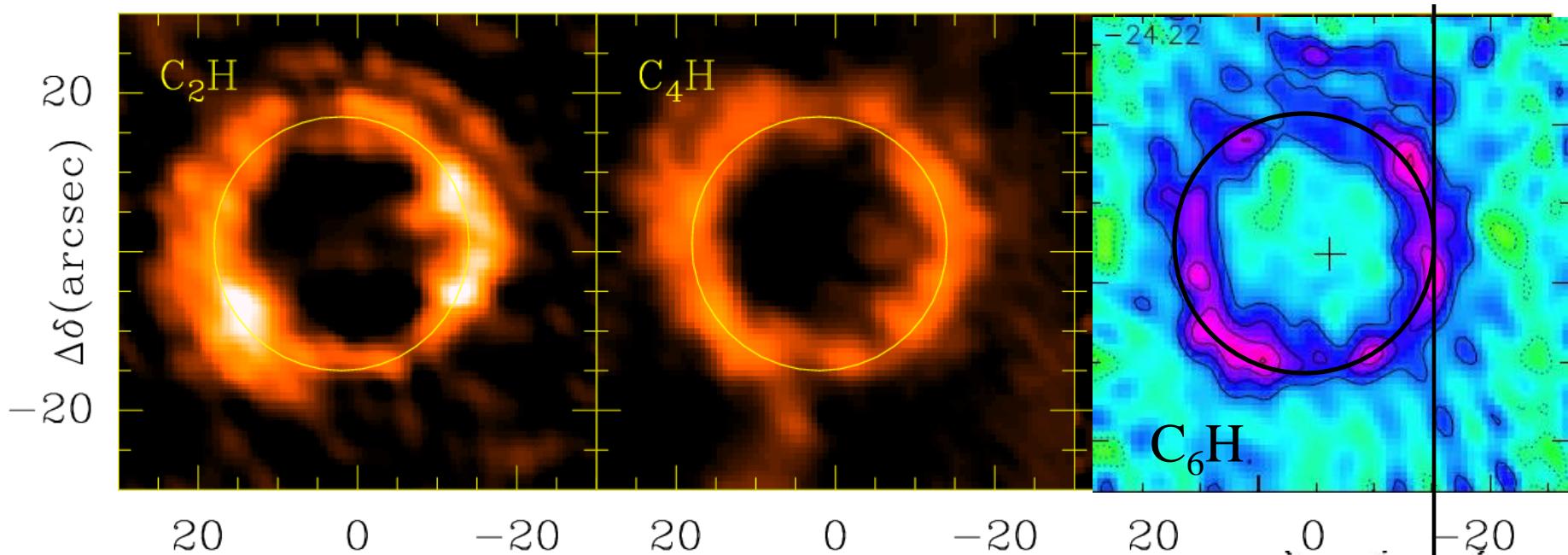


Plateau de Bure interferometer

Emission at $v = v_{\text{star}} \pm 3 \text{ km s}^{-1}$

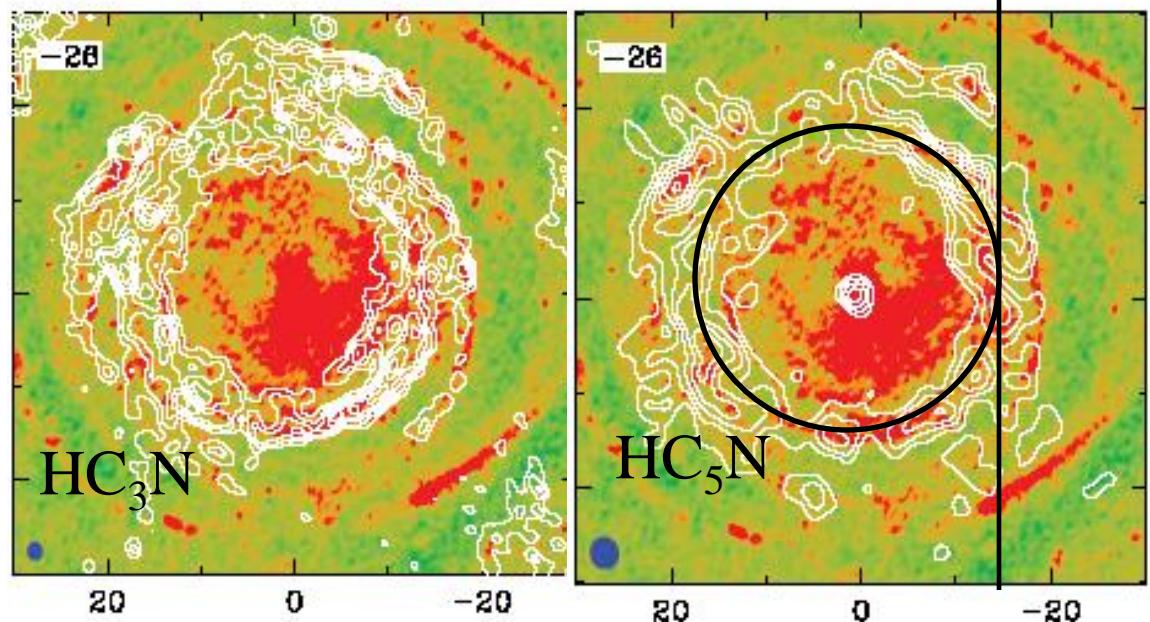


(Lucas & Guélin 1995)



Guélin et al. 2000

Emission at $v = v_{\text{star}} \pm 3 \text{ km s}^{-1}$

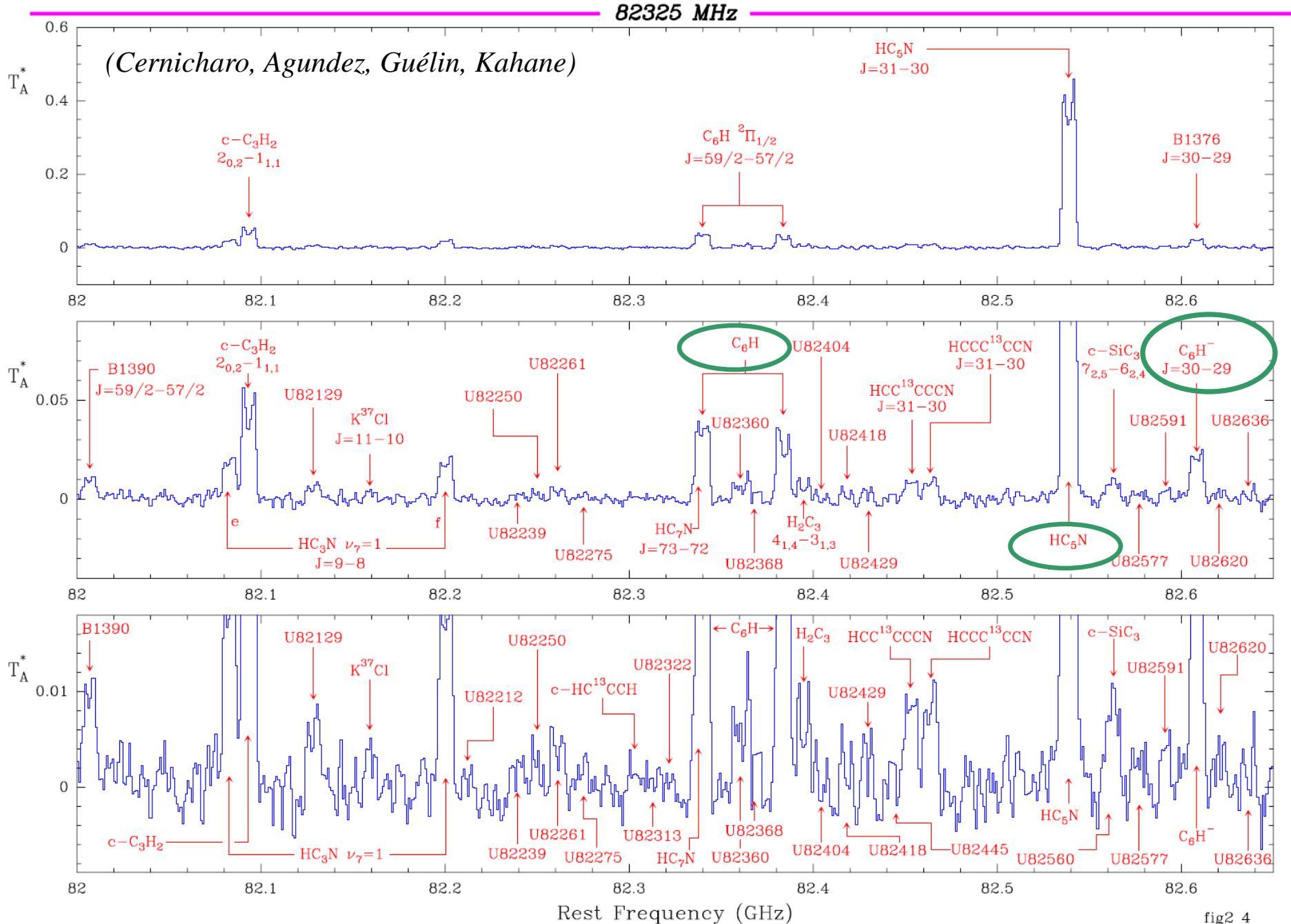


D-V-Trung & Lim 2006

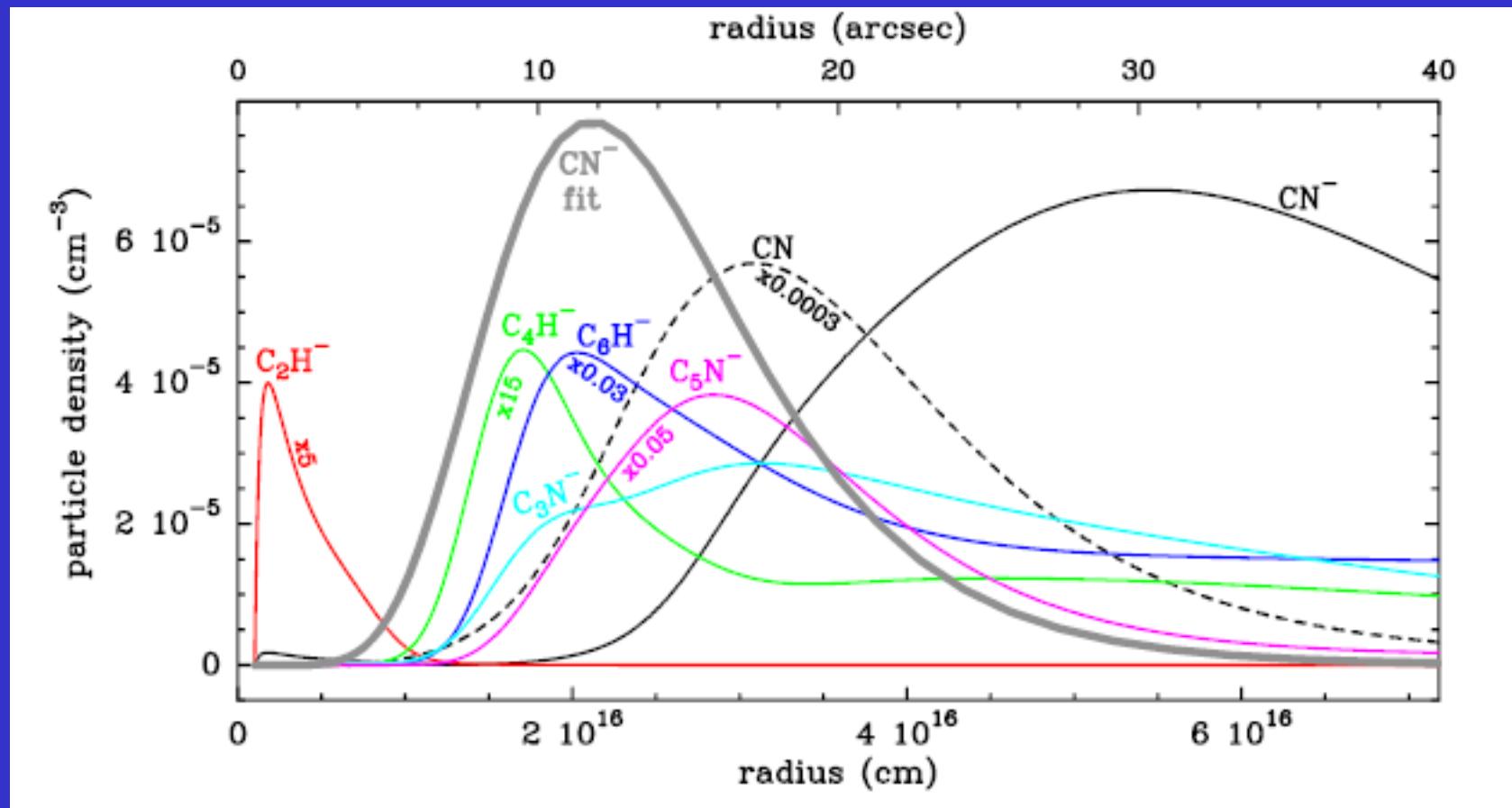
Observed interstellar anions (gas phase):

- C_2H^- ?? **IRC + 10216** (*one single line!*)
- C_4H^- **IRC + 10216, L1527**
- C_6H^- **IRC+10216, Dark Clouds TMC1, Lupus-1A , L1512 (Auriga), Protostellar cores L1527, L1544, L1251A (Cepheus)**
- C_8H^- **TMC1, Lupus-1A, IRC + 10216**
- C_3N^- **IRC + 10216**
- C_5N^- **IRC+10216** (*no lab data yet!*)
- CN^- **IRC+10216**

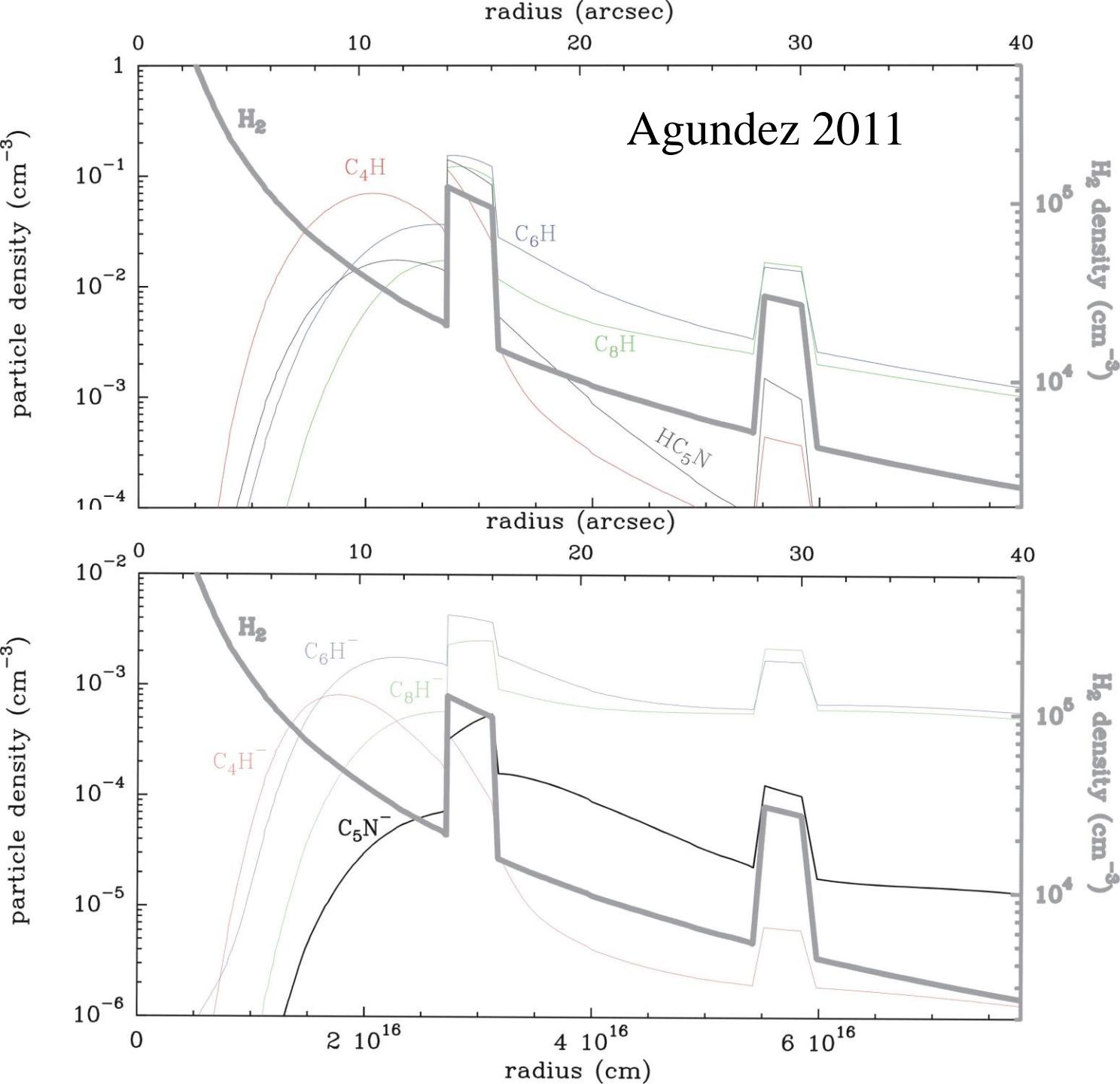
Excerpt from IRC+10216 spectral survey with the IRAM 30-m tel

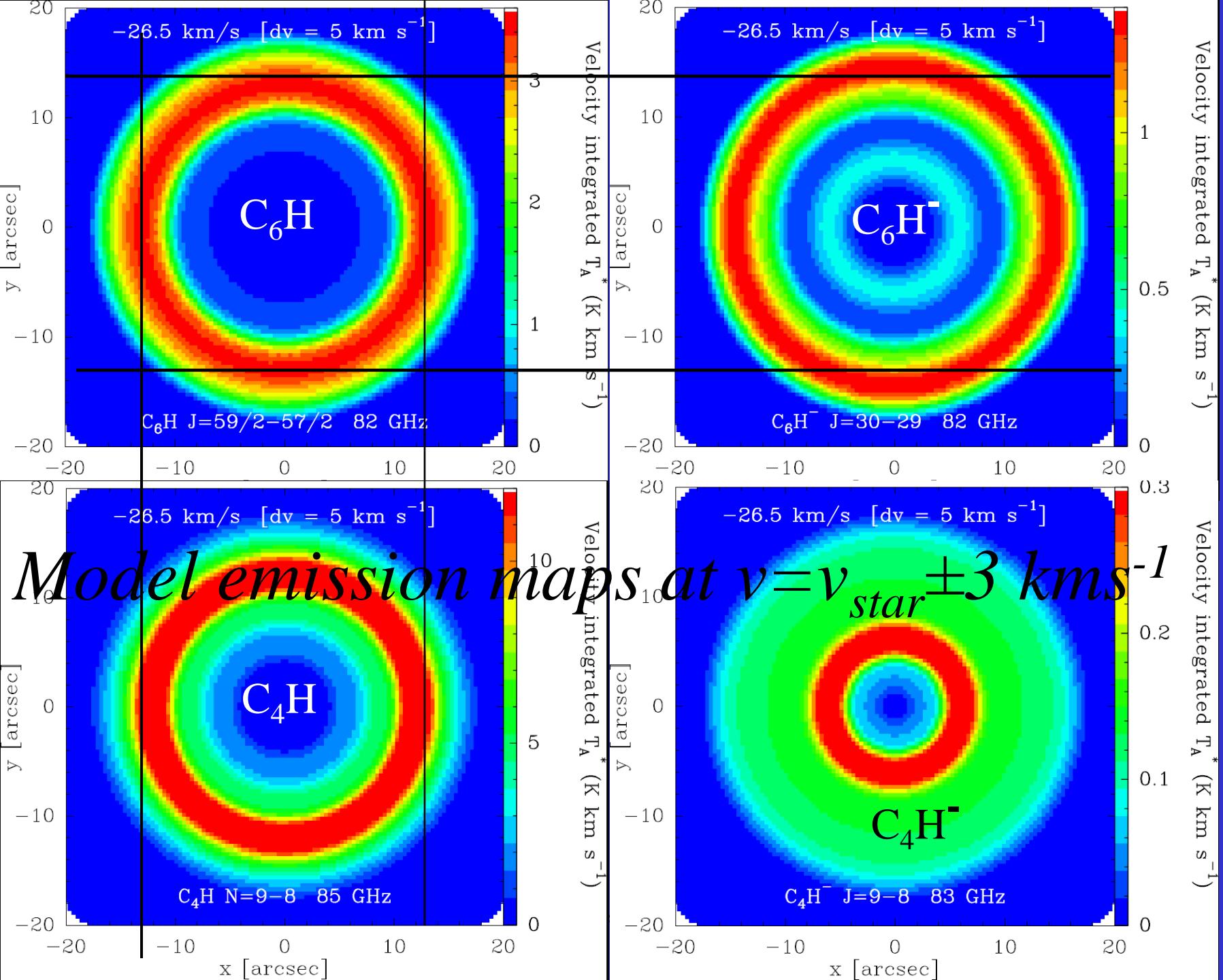


Expected model abundance of C-chain anions (smoothly decreasing density distribution)

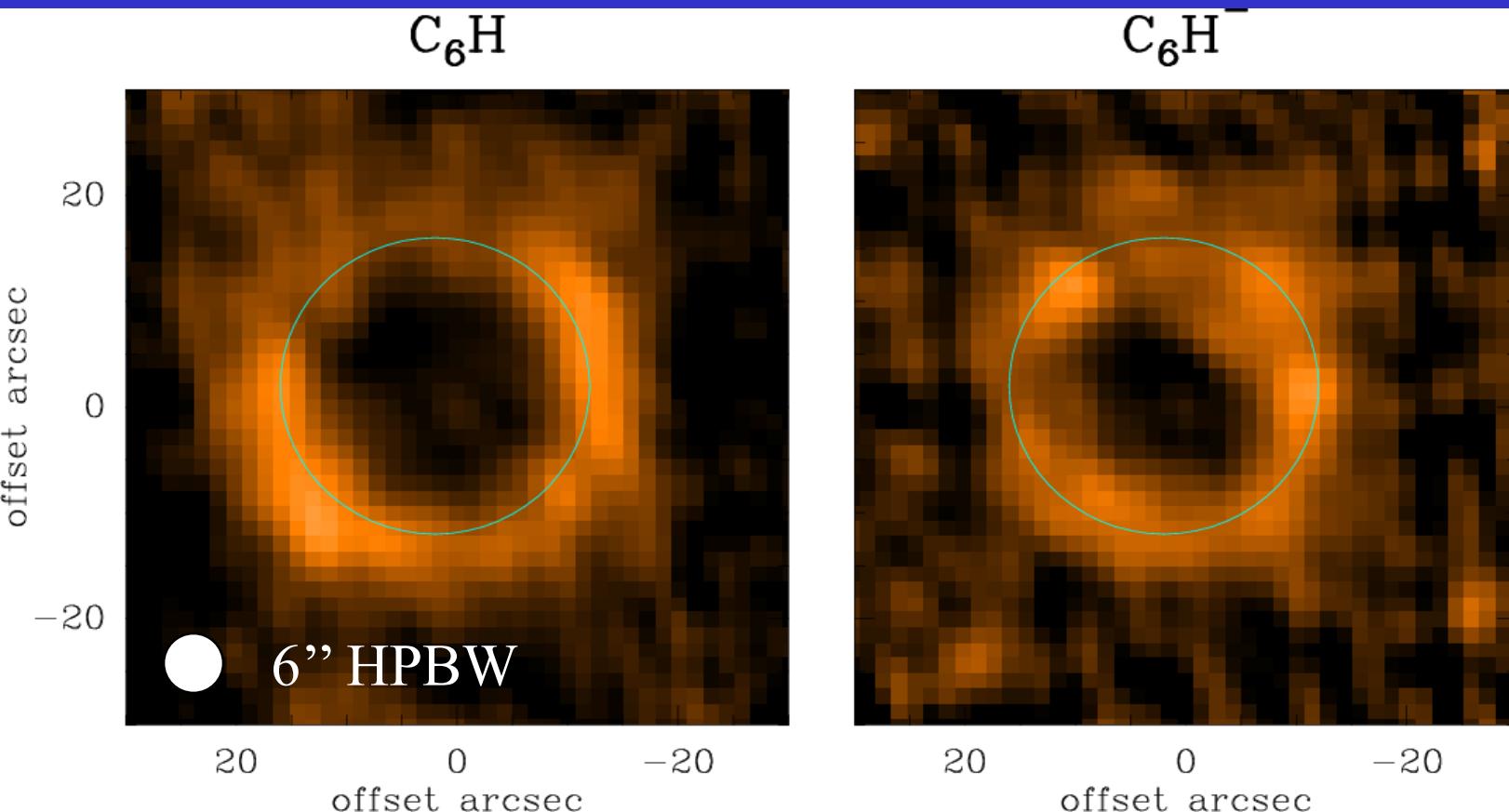


Expected molecular abundance for ringed density distribution

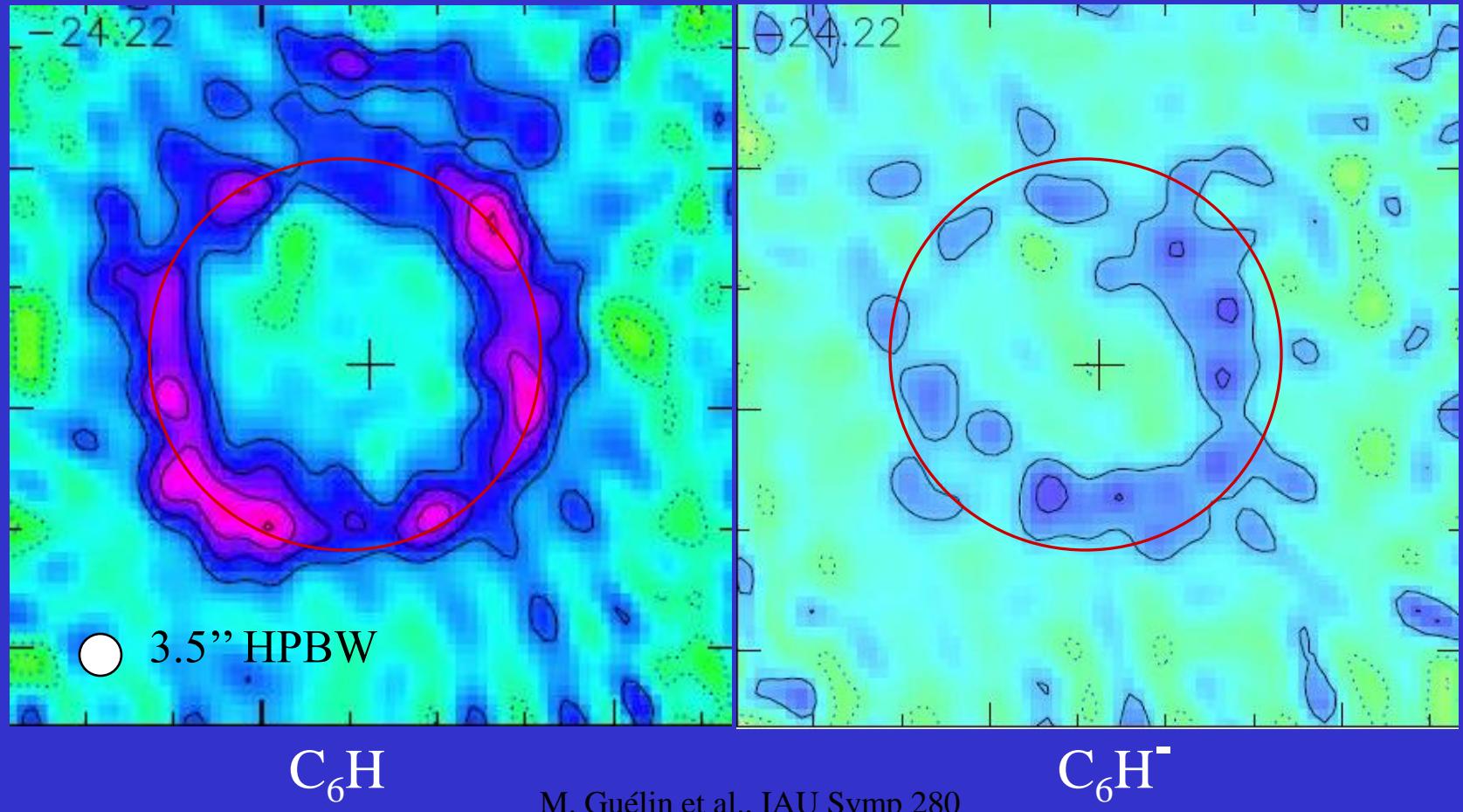




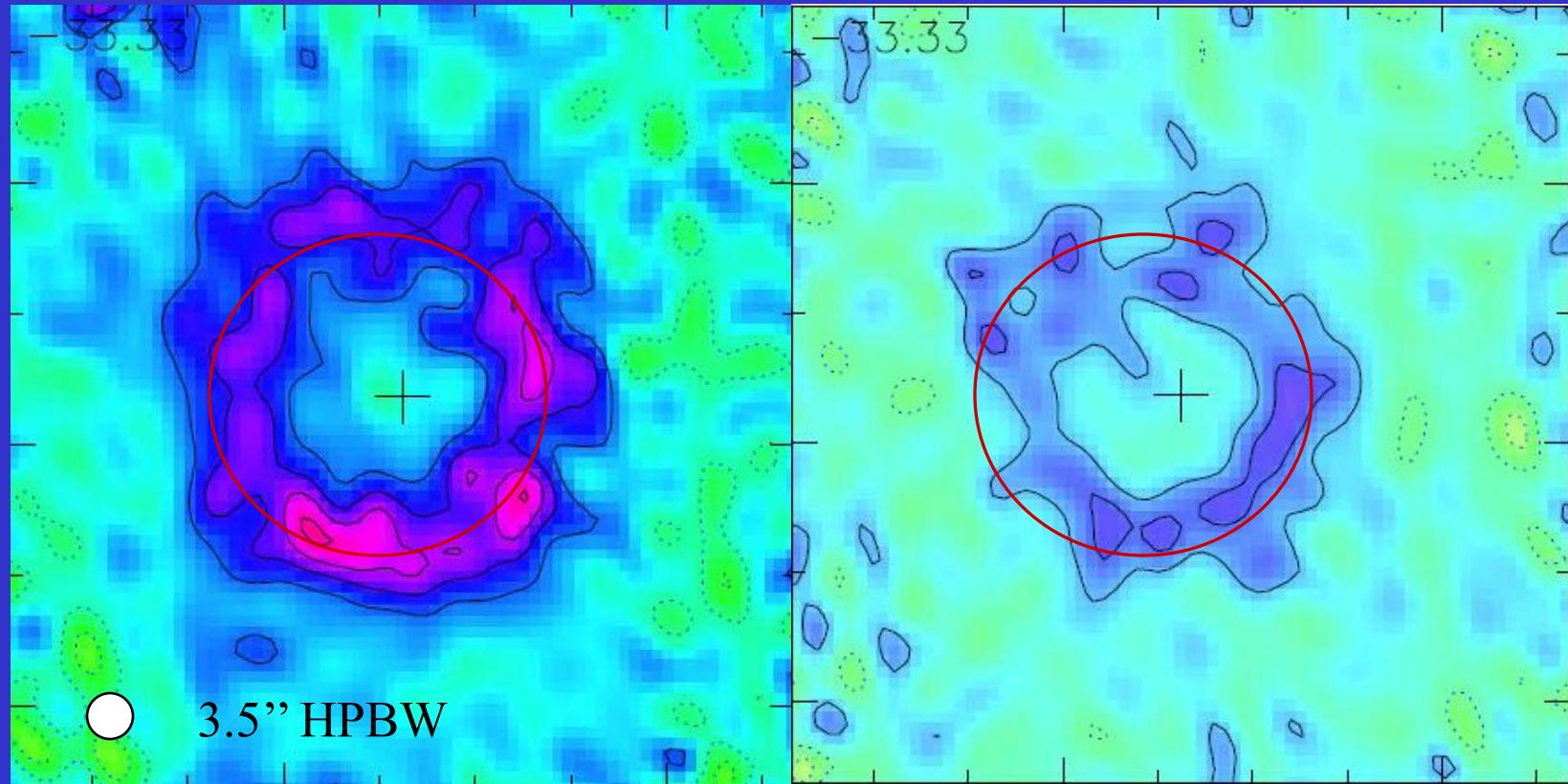
First map of an anion (*PdBI*, Guélin et al.)



New high angular&spectral resolution map of anions (*PdBI, Guélin et al.*)



New high angular&spectral resolution map of anions (*PdBI, Guélin et al.*)

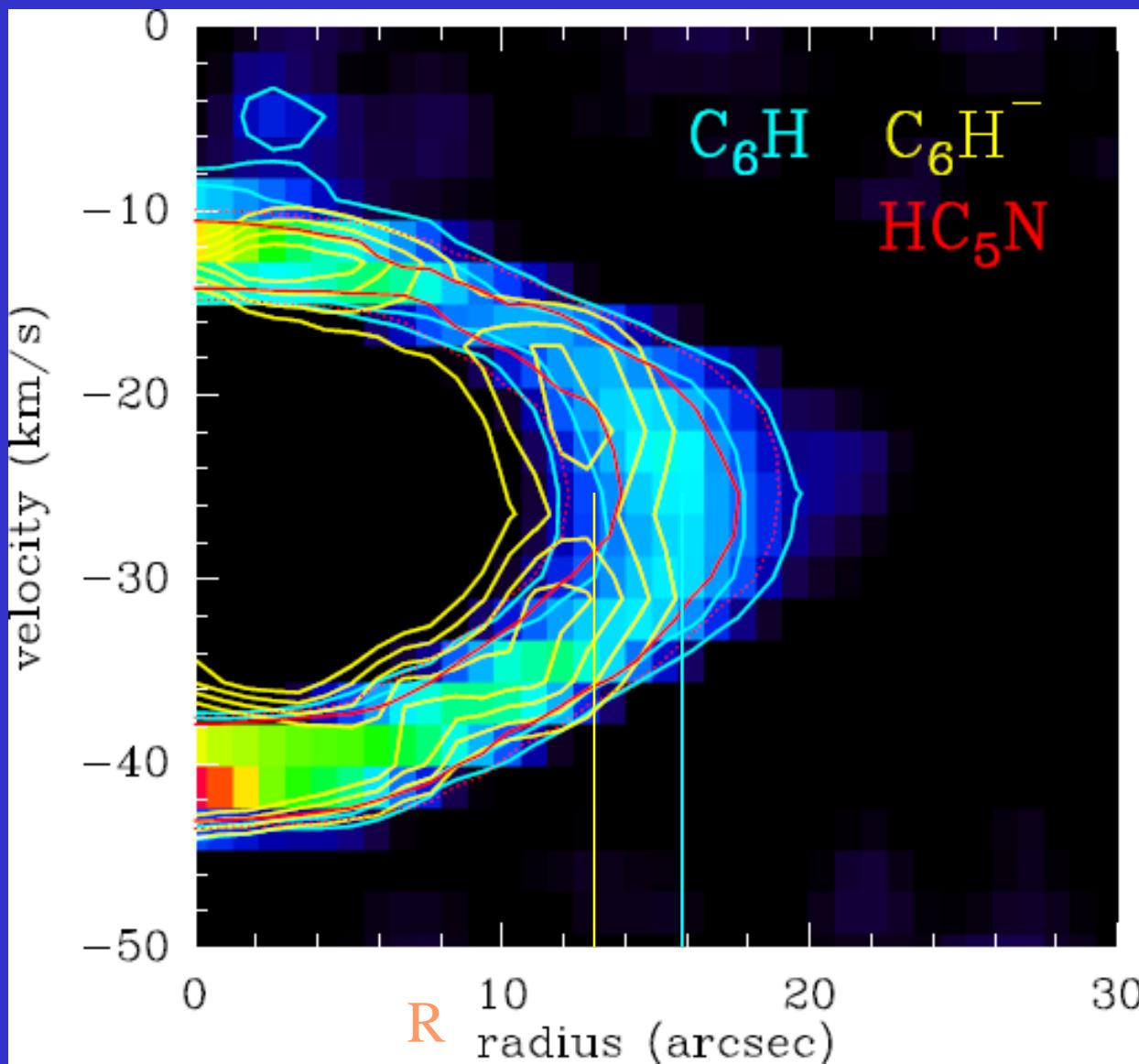


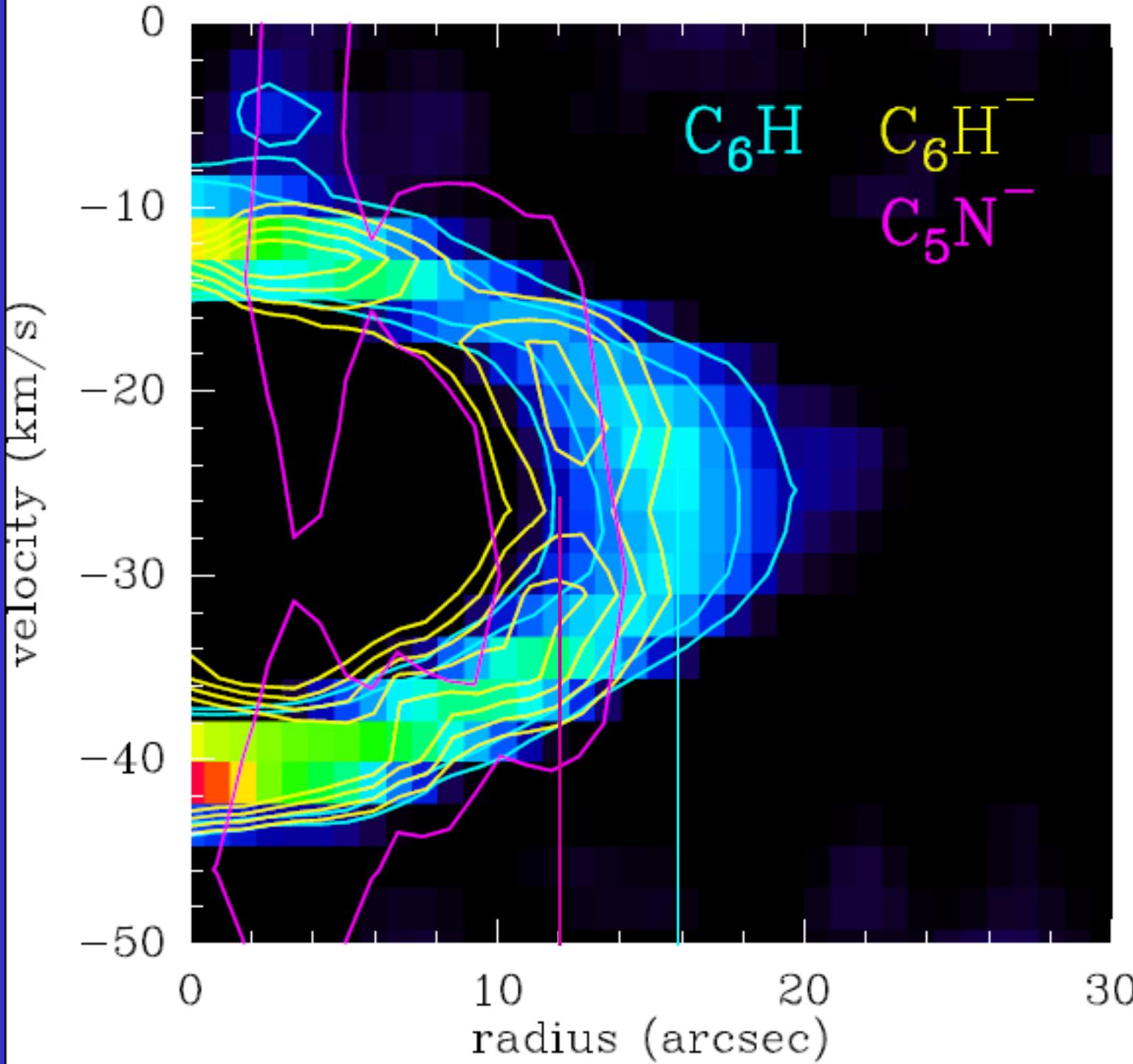
C_6H

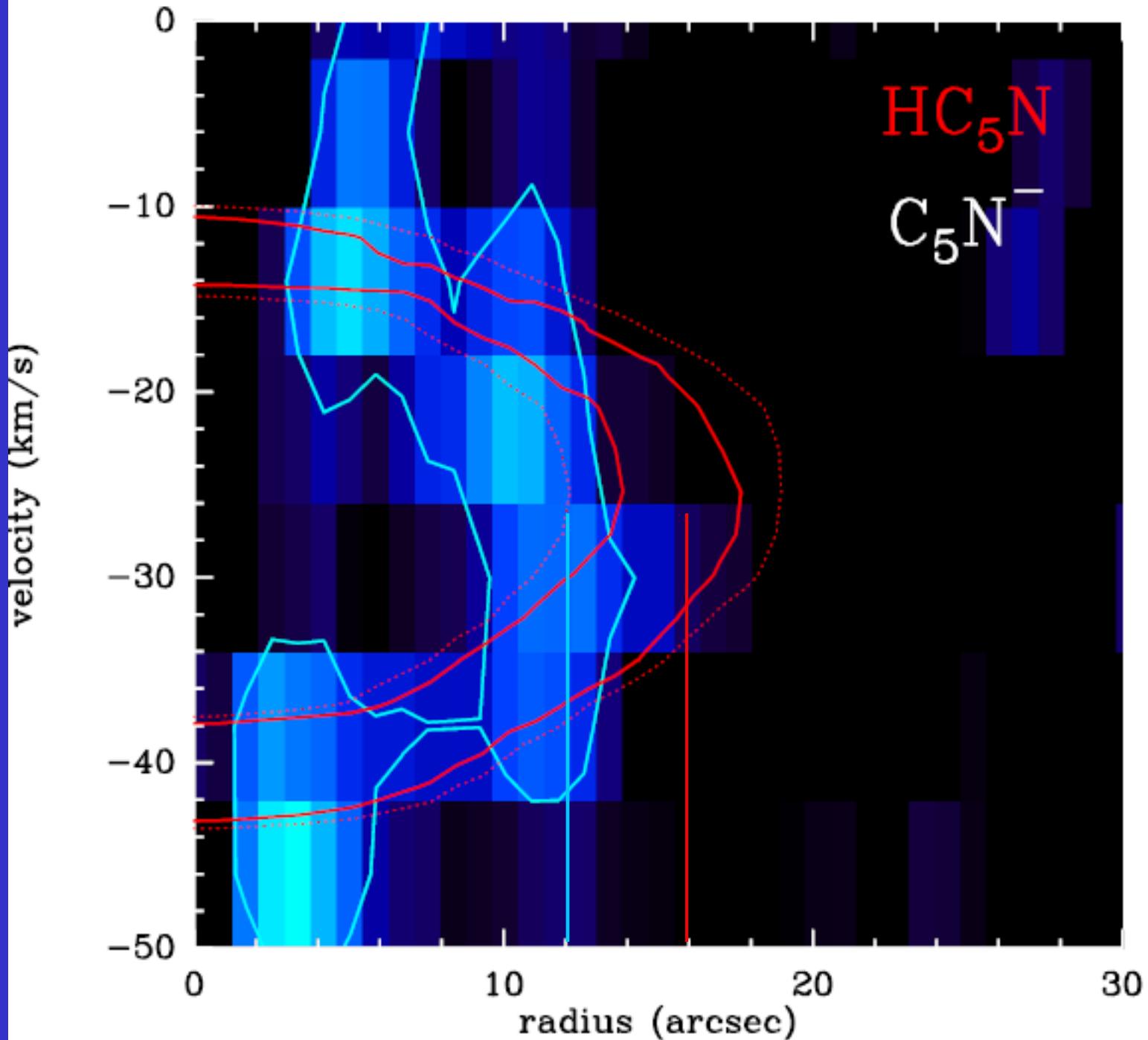
M. Guélin et al., IAU Symp 280

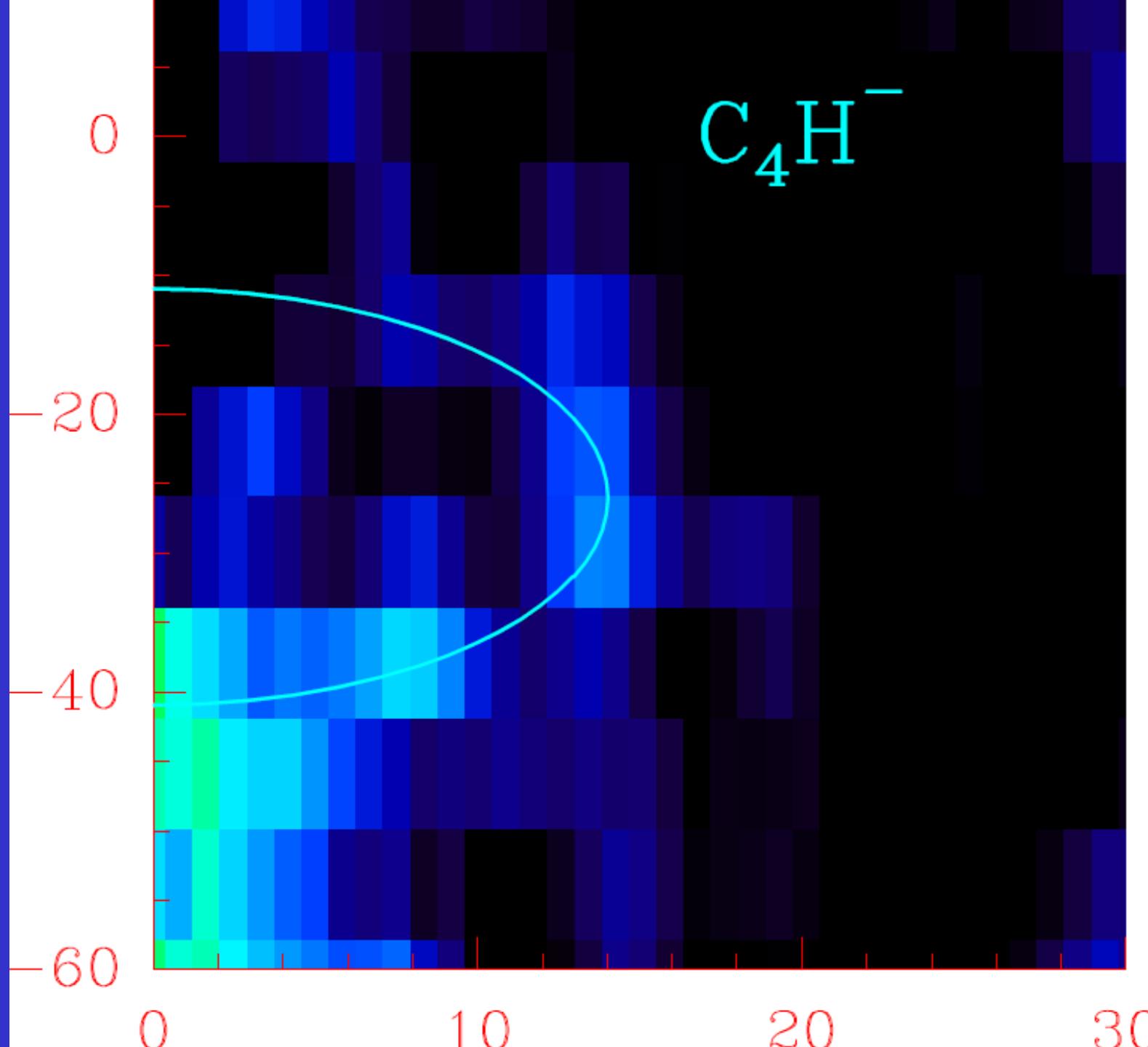
C_6H^-

Average intensity in concentric rings of radius R ,
for each velocity channel









First conclusion

C_6H^- and C_5N^- appear at smaller radii
(much earlier) than predicted.

C_4H^- appears further out (much later)
than predicted.

Neutral species	Activation Energy (eV)	Anion / Neutral (%)	rate * k_{ra} (astro) cm ³ s ⁻¹	rate ** k_{ra} (theor) cm ³ s ⁻¹
C ₂ H	3.0	<0.0014	< 10 ⁻¹¹	2.0 10 ⁻¹⁵
C ₄ H	3.6	0.0074	4 10 ⁻¹¹	1.1 10 ⁻⁸
C ₆ H	3.8	6.8	3 10 ⁻⁸	6.2 10 ⁻⁸
C ₈ H	4.0	26.	1.5 10 ⁻⁷	6.2 10 ⁻⁸
CN	3.8	0.25	2 10 ⁻⁹	1.4 10 ^{-17&}
C ₃ N	4.6	0.42	3 10 ⁻⁹	2 10 ⁻¹⁰ @
C ₅ N	4.5	58. (?)	5 10 ⁻⁷ (?)	

- * M. Agundez (PhD thesis 2009); rates scaled to 300 K
- ** Herbst & Osumara 2008, @Petrie & Herbst 1997
- @ Petrie 1996

Second conclusion

The observed abundances of C₄H- and CN- disagree with predictions based on direct electron attachment on C₄H and CN

Other formation mechanisms for CN⁻?

- $\text{CN} + \text{e} \rightarrow \text{CN}^-$
- $\text{MgNC} + \text{e} \rightarrow \text{CN}^-$
- $\text{C}_n^- + \text{N} \rightarrow \text{CN}^-$
- dust grains ?

Conclusions

- All 6 anions detected so far in gas phase have been seen in the **Circumstellar Envelope IRC+10216**
- Envelope expansion offers an unique opportunity to test the time-dependence of the chemistry
- Electron radiative attachment rates k_{ra} can be constrained
- There is definitively a problem with the theoretical estimates of $k_{ra}(\text{C}_4\text{H})$
- C_6H^- and C_5N^- appear at smaller radii (much earlier) than predicted. C_4H^- further out (much later)
- CN is probably mostly formed by another process: reaction of large C-anions (C_n^-) with N atoms, dissociative electron attachment on e.g. MgNC,..
- What's next? NCO- microwave spectrum recently observed at Harvard