The Chemistry of Exoplanetary Atmospheres

The Effects of Non-equilibrium Chemistry



Outline

Introduction

② Exoplanetary atmospheres

② Results

- Zero-metallicity
 C, N, O chemistry
- O Initial conditions
- ② Further work
- ② Summary



Transiting Exoplanets



Figure from Winn 2009, in Exoplanets, ed. S. Seager

'Hot Jupiters'

Ø Most transiting planets are 'Hot Jupiters'

- \odot Between < 1 and 10 M_i and < 1 and a few R_i
- Orbital radius of between 0.01 and 0.1 AU
- Orbital period on the order of 1 day
- Heavily irradiated by parent star
- Tidally locked with a constant night-side and dayside
- ② Low Bond albedo i.e. reflectivity

② Atmospheric chemical composition

- Likely molecular hydrogen and helium dominated
- Trace elements include C, N, O, S, metals
- Upper day-side atmosphere is heavily irradiated
- Heat redistribution from day-side to night-side
- O Atmospheric dynamics?

② Formation mechanisms

- ② Core accretion
- ② Disk fragmentation
- 2 Planetary migration

② Mass-loss from atmosphere

- ② Evaporation
- ② Roche-lobe overflow



Figure from Fortney et al. 2008, ApJ, 678, 1419

IAU 280 'The Molecular Universe'

Molecules in Exoplanet Atmospheres

Transmission Spectrum



Figure from Fortney et al. 2010, ApJ, 709, 1396

Emission Spectrum



Figure from Swain et al. 2009, ApJ, 690, L114

Current forward models struggle to reproduce observed spectra.

Molecules identified using backwards/fitting models. Many free parameters!

Non-equilibrium chemistry, non-LTE effects, temperature inversion?

Building a Model Atmosphere

Ommon assumptions

- ② Plane-parallel geometry
- ② Hydrostatic equilibrium
- ② Radiative equilibrium
- ② Chemical equilibrium
- ② Local thermodynamic equilibrium

② Model Ingredients

- ② Planetary system parameters
- ② Chemical elemental composition
- ② Absorption cross section data
- ② Reaction rate coefficients

Complications

- ② Convection
- ② Dynamics (e.g. winds)
- ② Mixing/diffusion
- ② Incomplete opacity line lists
- ② Photochemistry
- ② Clouds and dust



Exoplanet Atmosphere Model

Physics and chemistry of the atmosphere are intrinsically linked! Self-consistent solution is required!



7

Results: Zero Metallicity



- ② Model reproduces expected atmospheric structure
- Photochemistry and ion-molecule chemistry lead to chemical stratification and an ionised layer at high altitudes (similar to PDR/PPD)
- O Significant abundances of H⁻ and H₃⁺ both absorb at IR wavelengths!

Results: C, N, O Chemistry

- HD 189733 b
 - Ø M_p = 1.13 M_j
 - \bigcirc R_p = 1.14 R_J
 - ② T_{eff} = 4980 K (K1 − K2)
 - ② a = 0.031 AU
 - D = 19.3 pc
- Solar metallicity (Anders & Grevesse, 1989)
 - C = 2.455 x 10⁻⁴
 - N = 6.026 x 10⁻⁵
 - ② 0 = 4.571 x 10⁻⁴







Results: Initial Molecular Abundances

Use Link between protoplanetary disk chemical structure and planetary atmosphere composition?

② Gas-giant planet formation

- Coagulation of icy dust grains in outer disk --->protoplanet
- ② Accretion of remnant disk material to form gaseous envelope i.e. core accretion
- ② Planet migration from outer disk to inner disk
- Contamination of atmosphere?

② Generate initial abundances from a protoplanetary disk model?

Results: Initial Molecular Abundances



Crosses = gas-phase; Squares = grain-surface Walsh et al. 2010, 2011

② Planet formation site

Protoplanetary disk model

- **Output** Gas-phase chemistry
- ② Gas-grain interactions
- **③** Grain-surface chemistry
- Oxygen-rich low-metal initial elemental abundances

Results: Initial Molecular Abundances



Gas-phase initial abundances Solid lines = 5 AU; Dashed lines = 10 AU

- ② Major chemical reprocessing in the atmosphere
- Only minor differences in final chemical structure using disk gasphase initial molecular abundances



Grain-surface initial abundances Solid lines = 5 AU; Dashed lines = 10 AU

- ② Larger differences when using grainsurface initial molecular abundances!
- ④ H₂O becomes dominant molecule
- Order of magnitude differences in CO_2 , NH_3 and CO/CH_4 at 5 and 10 AU

Further Work

Immediate

- Investigate chemical feedback on atmospheric physical structure
- Addition of trace elements e.g. Na, Mg, Fe, Ti, V
- ⁽²⁾ Further exploration of initial conditions
- ② Generate synthetic emission and transmission spectra to compare with current and future observations

Use Future

- ② Mixing/diffusion
- Cloud formation, absorption and scattering
- Solid particles, aerosols and haze
- Super-Neptunes, Saturns and Earths
- Predictions for future missions e.g. JWST

Summary

Self-consistent model of an irradiated exoplanetary atmosphere with nonequilibrium chemistry

2 Zero-metallicity chemistry

- ② Chemical stratification in the atmosphere due to photo- and ion-molecule chemistry
- ② Absorption at IR wavelengths by H^- and H_3^+ ?
- ② Impact on physical structure?

Carbon, oxygen, nitrogen chemistry

- ② Solar metallicity
- ② Chemical stratification due to photo- and ion-molecule chemistry
- ② Atomic and molecular regions determined by H/H₂ boundary
- ② Dominant molecules in densest region: CO, CH₄, H₂O, NH₃
- ② Molecular absorption feedback?
- ② Varying metallicities?

② Initial molecular abundances

- ② Major chemical reprocessing in atmosphere
- ② Minor differences in final chemical structure using gas-phase initial abundances
- ② Larger differences in relative molecular abundances when using grain-surface/ice abundances
- ② Implications on planetary formation theory?